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Particulate removing catalyst filter and particulate removing method using the same.

The present invention relates to a particulate removing catalyst filter in which an oxidizing catalyst is carried on the surface portions of an airpermeable substrate through which the exhaust gas is adapted to be passed, and to a particulate removing catalyst filter in which a catalyst is carried on a thin wire net or metallic plate having a plurality of fine through-holes. The exhaust gas from a diesel engine is forcibly passed through the filter, thereby to remove particulates in the exhaust gas.

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Particulate Removing Catalyst Filter and Particulate Removing Method Using the Same

BACKGROUND OF THE INVENTION

The present invention relates to a particulate removing catalyst filter and a particulate removing method.
 5 More particularly, the present invention relates to a particulate removing catalyst filter for removing particulates contained in the exhaust gas from a combustion engine such as a diesel engine or the like, and to a particulate removing method using this catalyst filter.

The exhaust gas discharged from a diesel engine contains particulates. Such particulates contain not only soot (carbon), but also a variety of hydrocarbons from soft hydrocarbon to heavy hydrocarbon
 10 (polycyclic aromatic hydrocarbon), sulphuric acid mist (SO_4), and the like. There are some particulates of which particle sizes are as large as several thousand angstroms. However, it is generally considered that their particle sizes are in a range from about 100 to about 1000 angstroms with the average size being in a range from about 300 to about 500 angstroms.

Such particulates are produced due to incomplete combustion of diesel engine fuel or light oil. Recently, it has been strongly desired to prevent such particulates, as well as nitrogen oxide, from spreading in the atmosphere in view of prevention of air pollution.

In this connection, there have been proposed a variety of particulate removing catalyst devices arranged such that a catalyst filter carrying an oxidizing catalyst is disposed at the exhaust system of a diesel engine so that catalytic oxidation of the particulates occurs, causing the particulates to be finally
 20 decomposed and removed (for example, Japanese Publication of Unexamined Patent Application No. 185425/1988).

As a typical example of the catalyst filter of the type above-mentioned, there is known a catalyst filter formed in a honeycomb structure, in which the honeycomb structure substrate made of ceramic or the like, carries a catalyst component, or in which the honeycomb structure itself is made of a catalyst component.

However, the exhaust gas discharged from a diesel engine contains not only the particulates above-mentioned, but also gaseous components such as nitrogen oxide, gaseous hydrocarbon, CO, sulfur dioxide SO_2 and the like. Accordingly, when the particulates are oxidized and dissolved in the partition walls 2 of the honeycomb structure, the sulfur dioxide in the gaseous components is also oxidized and converted into sulfate by the same catalyst. Such sulfate may bring about acid rain or generate secondary particulates. In
 30 view of prevention of air pollution, it has been strongly desired to minimize the production of such sulfate.

On the other hand, while the engine is rotated at a low speed or with no load applied thereto, the exhaust gas temperature is lower (about 300°C) than that at a normal operation. This prevents the particulates from being sufficiently oxidized and dissolved by the catalyst filter. Accordingly, the particulates are accumulated in the partition walls 2 to prevent the exhaust gas from passing therethrough. This
 35 disadvantageously increases the pressure loss of the exhaust gas. If the pressure loss of the exhaust gas in the catalyst filter is increased, the fuel combustion condition in the diesel engine combustion chamber is deteriorated by the back pressure from the catalyst filter. This further accelerates the generation of the particulates, thereby to further increase the accumulation of the particulates on the partition walls 2. This results in increase in pressure loss. This involves the likelihood that the fuel combustion is incomplete in the engine.

To solve such a problem, the catalyst filter may be externally heated so that the oxidation and decomposition of the particulates sufficiently proceed even though the diesel engine is rotated at a low speed or with no load applied thereto. However, such arrangement requires an external heater and a heat sources therefor. This makes such arrangement impractical in view of the cost.

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SUMMARY OF THE INVENTION

50 It is an object of the present invention to provide a particulate removing catalyst filter capable of accelerating the decomposition of the particulates with the generation of sulfate restrained.

It is another object of the present invention to provide a particulate removing catalyst filter which assures high chances of the catalyst coming in contact with the particulates and in which there is no possibility of nonreacted particulates remaining, thereby to oxidize and remove the particulate in a stable manner for a long period of time without increase in pressure loss to increase the fuel cost, and also to

provide a particulate removing method having the filter above-mentioned.

The present invention provides a particulate removing catalyst filter in which an oxidizing catalyst is carried on the surface portions of an air-permeable substrate through which the exhaust gas is to pass. The term "in which an oxidizing catalyst is carried on the surface portions of a substrate" includes not only a mode in which the oxidizing catalyst is carried on only the surface portions of the substrate, but also a mode in which a major portion of the oxidizing catalyst is carried on at least the surface portions of the substrate.

According to the study of the Inventors, the particulate decomposition reaction velocity is faster than the oxidation speed of sulfur dioxide. Accordingly, in the particulate removing catalyst filter of the present invention in which the oxidizing catalyst is carried on the surface portions of the air-permeable substrate through which the exhaust gas is to pass, the particulates may be quickly oxidized and decomposed by the oxidizing catalyst on the surface portions, but the oxidation of sulfur dioxide does almost not proceed, by only the oxidizing catalyst on the surface portions, in such extent that the sulfur dioxide is converted into sulfate. This may restrain the generation of sulfate without preventing the removal of the particulates.

To enhance the particulate removing efficiency of such catalyst filter, there may be suitably applied, to the catalyst filter above-mentioned, the particulate removing method of the present invention by which the exhaust gas from a diesel engine is forcibly passed.

According to another embodiment of the present invention, there is proposed a particulate removing catalyst filter in which a catalyst is carried on a wire net or metallic plate having a plurality of fine holes.

The wire net or metallic plate above-mentioned may be mainly made of stainless steel.

In the catalyst filter, the exhaust gas is forcibly passed through the fine holes, assuring high chances of the particulates coming in contact with the catalyst. Further, the catalyst carrier is a thin wire net or metallic plate. This involves no likelihood that unreacted particulates remain inside, as done in a normal honeycomb structure filter. Accordingly, even in a long-term operation of a diesel engine or the like, the pressure loss in the catalyst filter is not increased, enabling the particulates to be removed in a stable manner.

To enhance the particulate removing efficiency of such catalyst filter, there may be suitably applied, to the catalyst filter above-mentioned, the particulate removing method of the present invention by which exhaust gas from a diesel engine is forcibly passed.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a section view of a catalyst filter in the form of a honeycomb structure in accordance with a preferred embodiment of the present invention;

Figure 2 is a section view of a catalyst device having the catalyst filter of the honeycomb structure in Figure 1;

Figure 3 is a plan view of flat and corrugated plates made of a ceramic fiber sheet;

Figure 4 is a perspective view of a honeycomb structure made of the corrugated and flat plates in Figure 3;

Figure 5 is a section view of a catalyst device having a catalyst filter made of a wire net or metallic plate;

Figure 6 is a section view of a catalyst device formed by overlapping, in the form of a coaxial cone, catalyst filters each of which is made of the wire net or metallic plate in Figure 5;

Figure 7 is a section view of a catalyst filter made of the wire net or metallic plate in Figure 5 which is secured, in a corrugated manner, between two large-mesh wire nets;

Figure 8 is a section view of a catalyst device formed by machining the catalyst filter in Figure 7, into a cylindrical shape;

Figure 9 is a section view of a catalyst device in which each of the catalyst filters made of the wire net or metallic plate above-mentioned is disposed as turned substantially at a right angle; and

Figure 10 is a section view of a catalyst device in which the catalyst filters made of the wire net or metallic plate above-mentioned are disposed in a zigzag manner.

DETAILED DESCRIPTION OF THE INVENTION

No particular restrictions are imposed on the catalyst filter in which an oxidizing catalyst is carried on the surface portions of an air-permeable substrate through which the exhaust gas is to pass, as far as the

exhaust gas can pass through the catalyst filter. For example, the catalyst filter may be used in the form of a honeycomb structure 1 as shown in Fig. 1. More specifically, the catalyst filter in Fig. 1 is formed such that the oxidizing catalyst is carried on the honeycomb structure 1 made of ceramic provided at the exhaust gas inlet side A with openings and at the exhaust gas outlet side B with openings, these openings being alternately closed. Partition walls 2 of the honeycomb structure 1 are made of porous ceramic. Accordingly, the exhaust gas (shown by arrows) introduced into the filter from the inlet side A is forcibly passed through the partition walls 2 and discharged through the outlet side B. The shapes of these openings are not particularly limited, but may be hexagonal, circular, rectangular, or triangular.

According to the present invention, the air-permeable substrate forms the partition walls 2 of the honeycomb structure 1, and may be suitably made of porous ceramic such as alumina, cordierite or the like, in view of its good particulate collecting efficiency and heat resistance. Since the air-permeable substrate is porous, a number of permeable fine pores are formed therein. The exhaust gas is adapted to pass through these permeable holes. The thickness of the air-permeable substrate is suitably determined, but is generally in a range from about 0.1 to about 1 mm.

The thickness from the surface of each surface portion carrying the catalyst is preferably in a range of, for example, 5 to 100 μm , preferably 10 to 50 μm , for a substrate having a thickness of 300 μm .

In order that the oxidizing catalyst is carried on the surface portions of the air-permeable substrate, the oxidizing catalyst may be carried directly on the substrate, or a catalyst carrier is previously carried, after which the oxidizing catalyst is then carried on this catalyst carrier. Alternatively, a carrier on which a catalyst component is previously carried, may be carried on the substrate. When a catalyst is to be carried directly on the substrate, the substrate surfaces may be previously treated such that the catalyst is suitably carried on only the surface portions of the substrate (for example, the substrate may be heated before the catalyst is carried thereon).

According to a more specific method of carrying the catalyst, the substrate may be immersed in a solution in which salt of the catalyst component has been dissolved, and then dried so that the catalyst component is carried on the surface portions of the substrate. Thereafter, the substrate may be calcined at a temperature of about 300 to 800 $^{\circ}\text{C}$. In order that the oxidizing catalyst is selectively carried on the surface portions of the substrate, the temperature raising speed at the time of drying may be adjusted. More specifically, when drying the substrate uniformly impregnated with the solution in which the salt of the catalyst component has been dissolved, a higher temperature raising speed causes the substrate to be dried with a more portion of the solution moved toward the surface portions of the substrate. Thus, the catalyst may be carried on the surface portions. However, if the temperature raising speed is excessively fast, the substrate is dried before the solution is moved. This fails to carry the oxidizing catalyst on only the surface portions of the substrate. It is therefore required to set a suitable temperature raising speed.

According to another method of carrying the catalyst on the surface portions of the substrate, a slurry of a carrier component (alumina powder or the like) may be first carried on the surface portions of the substrate, and the catalyst component may be then carried thereon. Alternatively, a slurry may be formed by carrying a catalyst component on a carrier component, and may be carried on the surface portions of the substrate.

According to the present invention, it is not always required that the oxidizing catalyst is carried on the surface portions of the substrate at both sides thereof, but the catalyst may be carried on either side of the substrate.

Examples of the oxidizing catalyst to be used in the present invention include a variety of conventional oxidizing catalysts including: a single metal such as platina, palladium; and manganese oxide, chromium oxide, copper chromate, iron oxide or the like.

Examples of the catalyst carrier to be used when carrying the catalyst on the substrate through a carrier, include alumina, titania, zirconia or the like.

For example, as shown in Fig. 2, the catalyst filter comprising the honeycomb structure 1 is disposed in a cylindrical case 3, thus forming a particulate removing catalyst device to be used as incorporated in the exhaust system of a diesel engine or the like.

According to the present invention, the catalyst filter may be used in the form of a honeycomb structure 6 shown in Fig. 4. This honeycomb structure 6 is formed by winding an assembly of a corrugated plate 4 overlapping on a flat plate 5 as shown in Fig. 3, both plates being made of a ceramic fiber sheet serving as the air-permeable substrate. The oxidizing catalyst is carried on the surfaces of the corrugated plate 4 and the flat plate 5.

The following description will discuss a catalyst filter in which the catalyst is carried on a thin wire net or metallic plate having a number of fine holes.

The wire net serving as the substrate is made with the use of a metallic wire member made of: a single

metal such as iron, cobalt, molybdenum, titanium, zirconium, chromium, silver, gold, copper, nickel, tin or the like; or an alloy such as an iron alloy including stainless steel, a copper alloy, a nickel alloy, a tin alloy, a chromium alloy and the like.

The metallic plate may be formed by piercing a number of fine holes, by etching or the like, in a thin metallic plate made of a metallic material similar to the metal wire material above-mentioned. In view of machinability, heat resistance and the like, stainless steel such as SUS 304 or the like may be preferably used as the metallic material forming the wire net or metallic plate. When the heat resistance is particularly required, there may be preferably used chrom-nickel steel, manganese-chrom steel, chrom-aluminium steel, nickel-chrom-cobalt steel or the like.

Each of the fine holes formed in the wire net or metallic plate (each of the meshes for the wire net) should have a diameter of 30 μm or more, preferably in a range from 100 to 500 μm , in view of the maximum grain size of the particulates and the pressure loss.

The thickness of the wire net or metallic plate is preferably in a range of from 0.05 to 0.5 mm in view of the strength of the catalyst filter and the passage of the particulates.

The wire net or metallic plate having the arrangement above-mentioned is extremely thin and has fine holes having diameters which do not prevent the particulates from passing therethrough. This involves no likelihood that unreacted particulates remain in the catalyst.

For carrying the catalyst on the substrate, there are available four methods as outlined below.

According to a first method, a catalyst carrier component such as alumina or the like is spray-coated on the wire net or metallic plate, on which the catalyst is then carried.

When there is used the wire net or metallic plate made of iron such as stainless steel or the like, there may be used a second, third or fourth method.

According to the second method, this iron wire net or metallic plate is heated so that the surface thereof is oxidized with the air to form an iron oxide layer on the surface. On this iron oxide layer, the catalyst is carried. According to the third method, the iron constituting the wire net or metallic plate is partially eluted, causing the wire net or metallic plate to be porous. A catalyst carrier such as alumina or the like is carried on the porous wire net or metallic plate, on which the catalyst is then carried. According to the fourth method, a portion of the iron constituting the wire net or metallic plate is substituted with active platinum, palladium or the like, as a catalyst. The iron oxide above-mentioned may be used as the catalyst for oxidizing and decomposing the particulates. Accordingly, the method of merely heating the wire net or metallic plate made of iron such as stainless steel or the like so that the surface layer thereof is oxidized with the air, is included in the method of carrying the catalyst on iron oxide on the substrate.

As the catalyst to be carried on the substrate, there may be used any of the examples of the catalyst above-mentioned.

The catalyst filter may be shaped in a desired form or may have a desired arrangement by machining or combining the wire net or metallic thin plate carrying the catalyst thereon. With the use of such filter, the catalyst device may be shaped in different forms and arranged in different structures.

The following description will discuss embodiments of the catalyst device having the catalyst filter above-mentioned, with reference to the drawings.

In Fig. 5, a catalyst device 7 includes a cylindrical casing 8 of which both ends are opened, and a plurality of catalyst filters 9 made of a wire net carrying a catalyst. The catalyst filters 9 are disposed in the casing 8 in parallel with one another at spatial intervals. When this catalyst device 7 is installed in the exhaust gas system of a diesel engine and the exhaust gas is forcibly passed through the catalyst filters 9 as shown by arrows in Fig. 5, the particulates in the exhaust gas are securely removed by a plurality of the catalyst filters 9.

Fig. 6 shows another embodiment of the catalyst device 7, in which a plurality of catalyst filters 9 are overlapped in the form of a coaxial cone. This catalyst device 7 is opened at the exhaust gas inlet side thereof, and closed at the rear end thereof by a wall 10 which prevents the exhaust gas from passing therethrough. Thus, the exhaust gas is forcibly passed through the catalyst filters 9 to oxidize the particulates. A plurality of such catalyst devices may be disposed in series in the exhaust gas system as necessary. The arrangement in which a plurality of conical catalyst filters 9 are overlapped, presents the advantage that the contact area per catalyst capacity is increased without increase in pressure loss.

As shown in Fig. 7, a catalyst filter 9 may be secured in a corrugated manner between two support members, for example large-mesh wire nets 11, permitting the exhaust gas to easily pass therethrough. This assembly is machined in the form of a cylinder of which rear end is closed by a wall 10 which prevents the exhaust gas from passing therethrough, as shown in Fig. 8. This cylindrical assembly is mounted on a suitable casing (not shown), thus forming a catalyst device. In this catalyst device, the exhaust gas introduced therein is forcibly passed through the catalyst filter 9. A plurality of such catalyst

devices may be coaxially disposed as necessary. The arrangement in which the catalyst filter 9 is formed in the form of a corrugation, presents the advantage that the contact area per catalyst capacity is further increased as compared with the arrangement shown in Fig. 6.

In these embodiments of the catalyst device, the shape of the catalyst filter, the exhaust gas flowing angle and the like may be changed in different manners as necessary. For example, as shown in Fig. 9, the particulate removing catalyst filters 9 may be each turned substantially at a right angle and disposed in a casing 8 in parallel with one another at spatial intervals. Alternatively, as shown in Fig. 10, a plurality of particulate removing catalyst filters 9 may be installed in a casing 8 in a zigzag manner. Further, a long particulate removing catalyst filter 9 as turned in a zigzag manner may be installed in the casing 8. In these embodiments, the contact areas of the catalyst filters 9 housed in the casing 8 with the exhaust gas, are increased, thereby to improve the exhaust gas treating efficiency.

EXAMPLES

The following description will discuss, in more detail, examples of the present inventions, but the present invention should not be limited to these examples only.

Example 1

[Preparation of a carrier in the form of a honeycomb structure]

10 Kgs. of active alumina (A-11 manufactured by Sumitomo Kagaku Kogyo Co., Ltd.), 1 kg of kibushi clay and 500 grs. of methyl cellulose were mixed in a dry condition. The resultant mixture with water added thereto was sufficiently kneaded and extruded from an auger-type extruder on which there was mounted a die having a pitch of 1.33 mm and a wall thickness of 0.3 mm. Thus, a honeycomb was extrudingly formed. The honeycomb was dried with a ventilation-type drier, after which the temperature of the honeycomb was raised at a ratio of 5°C per one hour. The honeycomb was calcined at 500°C for one hour, thus forming a honeycomb structure having a diameter of 190 mm and a length of 150 mm.

[Preparation of a platina carrying catalyst]

The honeycomb structure above-mentioned was immersed in a 25 g/l chloroplatinic acid aqueous solution and then dried with hot air at 110°C for one hour with a circulation-type drier. At this time, the temperature of the honeycomb structure was raised to 110°C within 10 minutes. Then, the honeycomb structure was calcined at 500°C for one hour, thus forming an alumina-platina carrying honeycomb structure in which 0.1 % by weight of platina with respect to the alumina was carried.

As shown in Fig. 1, the opening ends at the exhaust gas inlet side and the opening ends at the outlet side of the honeycomb structure were alternately closed, thus forming a catalyst filter of the type in which the exhaust gas is forcibly passed through the partition walls.

[Preparation of a particulate removing catalyst device]

As shown in Fig. 2, the catalyst carrying honeycomb structure 1 thus obtained, was installed in a cylindrical case 3, thus forming a particulate removing catalyst device.

Example 2

A particulate removing catalyst device was obtained in the same manner as Example 1, except that the temperature of the honeycomb structure was raised to 110°C within 15 minutes in the preparation of the platina carrying catalyst.

Example 3

A particulate removing catalyst device was obtained in the same manner as Example 1, except that the temperature of the honeycomb structure was raised to 110°C within 20 minutes in the preparation of the platina carrying catalyst.

Example 4

A particulate removing catalyst device was obtained in the same manner as Example 1, except that the temperature of the honeycomb structure was raised to 110°C within 25 minutes in the preparation of the platina carrying catalyst.

Example 5

A particulate removing catalyst device was obtained in the same manner as Example 1, except that the temperature of the honeycomb structure was raised to 110°C within 5 minutes in the preparation of the platina carrying catalyst.

Example 6

In the honeycomb structure obtained in [Preparation of a carrier in the form of a honeycomb structure] in Example 1, the opening ends at the exhaust gas inlet side and the opening ends at the outlet side were alternately closed. Then, an alumina slurry was poured into the openings at the exhaust gas inlet side. After the excessive alumina had been removed, the honeycomb structure was dried and then calcined at 500°C for one hour, so that alumina was carried on only one sides of the partition walls 2. Then, a 25g/l chloroplatinic acid aqueous solution was poured into the same openings. The honeycomb structure was dried with hot air at 110°C for one hour with the use of a circulation-type drier. The temperature of the honeycomb structure was raised to 110°C within 30 minutes. Then, the honeycomb structure was calcined at 500°C for one hour, thus forming an alumina-platina carrying honeycomb structure in which 0.1 % by weight of platina with respect to the alumina was carried.

This honeycomb structure is a catalyst filter in which the catalyst was carried on the surfaces at only one side of the partition walls, i.e., the exhaust gas inlet side.

Thereafter, a particulate removing catalyst device was prepared in the same manner as Example 1.

Comparative Example 1

A particulate removing catalyst device was obtained in the same manner as Example 1, except that the temperature of the honeycomb structure was raised to 110°C within 120 minutes in the preparation of a platina carrying catalyst.

Example 7

A ceramic fiber sheet having a thickness of 0.3 mm (#2813 manufactured by Nichiasu Co., Ltd.) was cut into pieces each having a width of 150 mm, from which a corrugated plate 4 and a flat plate 5 were made as shown in Fig. 3. These plates 4, 5 were overlapped to form an assembly with the height h of each mountain portion being 1.7 mm and the width w thereof being 3.0 mm. As shown in Fig. 4, this assembly was wound to form a honeycomb structure 6 having a diameter of 190 mm and a length of 150 mm.

This honeycomb structure was immersed in alumina slurry identical with that in Example 1. After the excessive alumina had been removed, the honeycomb structure was dried and then calcined at 500°C for one hour so that the alumina was carried. The honeycomb structure was further immersed in a 25g/l chloroplatinic acid aqueous solution, and then dried with hot air at 110°C for one hour with the use of a circulation-type drier. The temperature of the honeycomb structure was raised to 110°C within 30 minutes. Then, the honeycomb structure was calcined at 500°C for one hour, thus forming an alumina-platina

carrying honeycomb structure in which 0.1 % by weight of platina with respect to the alumina was carried.
Thereafter, a particulate removing catalyst device was formed in the same manner as Example 1.

5. Particulate Removing Test

The particulate removing catalyst device thus obtained, was installed downstream of the exhaust system of a diesel engine presenting an exhaust amount of 12 liters and an exhaust gas amount of 700Nm³/hour. A test of particulate oxidation was conducted on this diesel engine. In the test, the diesel engine was operated at 2000 rpm under torque of 100 kgs m. Under these operating conditions, the compositions of the engine exhaust gas were as follows:

Component	Concentration
NO	1500 ppm
SO ₂	150 ppm
O ₂	5 %
H ₂ O	10 %

The average concentration of the particulates in the exhaust gas was equal to 1.0 g/Nm³. The reaction temperature was adjusted by externally heating the honeycomb structure.

With the diesel engine operated under the conditions above-mentioned, the particulate removing ratio was obtained. More specifically, there was used a so-called dilution tunnel method in which the exhaust gas from the diesel engine was diluted to generate a situation similar to that in which the exhaust gas was actually discharged into the atmosphere, and the amount of particulates was measured in the situation thus generated. According to this method, there was measured the weight of fine particulates collected on a filter having a thickness of 47μm mounted on the particulate removing catalyst device at the outlet side. Based on the weight data thus measured, there were obtained the removal ratio of soluble organic components (SOF) and insoluble organic components (IOF), the sulfate generation ratio and the overall particulate removing ratio. As to the SOF and IOF amount, the SOF and IOF were subjected to quantitative analysis with a liquid chromatography after dissolved in an organic solvent. The amount of hydrochloride was measured by a barium-thorin photometric titration method. Further, portions of the partition walls of the catalyst carrying honeycomb structures obtained in Examples 1 to 7 and Comparative Example 1, were cut out. The amounts of catalyst (Pt) in the portions thus cut, were measured with an X-ray micro-analyzer. The relationships between the distances from the wall surfaces wherein the catalyst was immersed, and the catalyst amounts were obtained. Then, there were set, as the thicknesses of the catalyst layers, the depths of the wall parts in which the ratios of the catalyst amounts to the entire carried catalyst amounts, exceed 95%. In Examples 1 to 5, and Comparative Example 1, each thickness value represents the total sum of the catalyst layer thicknesses of both sides of each partition wall. The test results are shown in Table 1.

Table 1

Examples							Comparative Example
1	2	3	4	5	6	7	1
Thickness of Catalyst Layer (μm)							
5	12	36	38	56	23	28	300*
SOF & IOF Removing Ratio (%)							
62.3	73.8	76.1	74.0	75.2	78.1	75.5	80.5
Sulfate Generation Ratio (%)							
0.3	0.2	0.5	3.1	16.8	0.2	0.8	389.4
Overall Particulate Removing Ratio (%)							
62.1	74.2	75.6	71.8	55.3	78.1	74.5	-308.9

* The catalyst layer was carried uniformly on the entire partition wall.

As apparent from Table 1, in Examples 1 to 7, the amounts of sulfate produced due to the oxidation of sulfur dioxide are small and, therefore, the SOF and IOF removing ratios represent, as they are, the particulate removing ratios. In Comparative Example 1, the SOF and IOF are decomposed, but the ratio of generating sulfate which newly constitutes particulates, is high. As the result, the particulates are increased in amount.

Example 8

[Preparation of a platina carrying wire net]

A wire net of SUS 304 having a diameter of 35 μm and meshes of 44 μm , was cut into a circular disk. After degreased at 400°C for 30 minutes, this disk was immersed in a 25g/l chloroplatinic acid aqueous solution for one minute. This disk was then dried, thus forming a platina carrying wire net with 0.05 gr. of platina carried on the wire net surface.

[Preparation of a particulate removing catalyst device]

Twenty five catalyst carrying wire nets each obtained in the manner above-mentioned, were disposed in a cylindrical casing at spatial intervals of 5 mm, as shown in Fig. 5, thus forming a particulate removing catalyst device.

Example 9

[Preparation of an alumina-platina carrying wire net]

After 50 grs. of alumina had been spray-coated on a wire net identical with that in Example 8, the wire net was immersed in a 25g/l chloroplatinic acid aqueous solution for one minute, and then dried. The wire

net was calcined at 500° C for one hour, and then reduced in a current of a hydrogen-nitrogen mixture at 400° C for one hour, thus forming an alumina-platina carrying wire net in which 0.1 % by weight of platina with respect to the alumina was carried.

Thereafter, a particulate removing catalyst device was obtained in the same manner as Example 8.

Example 10

10 [Preparation of a CuO-Cr₂O₃-MnO₂ carrying wire net]

After 50 grs. of alumina had been spray-coated on a wire net identical with that of Example 8, the wire net was successively immersed in aqueous solutions of copper nitrate, chromium nitrate and manganese nitrate. The wire net was then calcined at 500° C for one hour, thus forming an alumina-metal oxide carrying wire net in which the total amount of the metal oxide with respect to alumina was 5 % by weight, the metal oxide containing CuO/Cr₂O₃/MnO₂ at a ratio by weight of 37.5/36/1.8.

Thereafter, a particulate removing catalyst device was obtained in the same manner as Example 8.

20 Example 11

[Preparation of an alumina-palladium carrying wire net]

After 50 grs. of alumina had been spray-coated on a wire net identical with that of Example 8, the wire net was immersed in a 120 g/l palladium-chloride aqueous solution, and then dried. The wire net was then calcined at 500° C for one hour, and reduced in a current of a hydrogen-nitrogen mixture at 400° C for one hour, thus forming an alumina-palladium carrying wire net in which 1 % by weight of palladium with respect to the alumina was carried.

Thereafter, a particulate removing catalyst device was obtained in the same manner as Example 8.

Comparative Example 2

A ceramic foam having a diameter of 150 mm and a thickness of 30 mm (Ceramic Foam #20 manufactured by Bridgestone Co., Ltd., which has a hole ratio of 87.5% and is made of cordierite and alumina) was immersed in an alumina slurry. After the excessive alumina had been removed, the wire net was dried. Then, the wire net was calcined at 500° C for one hour so that 674 grs. of alumina was carried. The wire net was further immersed in a 20g/l chloroplatinic acid aqueous solution, and then dried. The wire net was calcined at 500° C for one hour, and then reduced in a current of a hydrogen-nitrogen mixture at 400° C for one hour, thus forming an alumina-platina carrying ceramic foam in which 0.1 % by weight of platina with respect to the alumina was carried.

Two catalyst carrying ceramic foams each obtained in the manner above-mentioned, were overlapped, and then disposed in a cylindrical casing similar to the casing shown in Fig. 5, thus forming a particulate removing catalyst device.

Comparative Example 3

A ceramic fiber sheet having a diameter of 150 mm and a length of 300 mm (#2813 manufactured by Nichiasu Co., Ltd.) was immersed in an alumina slurry. After the excessive alumina had been removed, the sheet was dried and then calcined at 500° C for one hour, so that 1060 grs. of alumina was carried. This sheet was further immersed in a 25g/l chloroplatinic acid aqueous solution, and then dried. The sheet was calcined at 500° C for one hour, and then reduced in a current of a hydrogen-nitrogen mixture at 400° C for one hour, thus forming a honeycomb structure of an alumina-platina carrying ceramic fiber in which 0.1 % by weight of platina with respect to the alumina was carried. In the honeycomb structure thus obtained, half the number of honeycomb openings at the exhaust gas inlet side was closed while the remaining half the number of honeycomb openings at the exhaust gas outlet side was closed, thus forming a catalyst filter of

the type in which the exhaust gas is forcibly passed through the honeycomb walls.

This catalyst filter was disposed in a cylindrical casing similar to that shown in Fig. 5, thus forming a particulate removing catalyst device.

With the use of each of the particulate removing catalyst devices of Examples and Comparative Examples, a diesel engine was operated under the conditions above-mentioned. There were measured particulate removing ratios and pressure loss data immediately after, one hour after and 24 hours after start of the operation, respectively. The pressure loss data were obtained by measuring the static pressures between the inlet and outlet sides of the particulate removing catalyst devices. According to the dilution tunnel method, there were measured the weights of fine particulates collected on a $47\mu\text{m}$ -filter disposed at the outlet sides of the particulate removing catalyst devices. Based on the weight data thus obtained serving as particulate amounts, the particulate removing ratios were obtained. The results are shown in Table 2.

Table 2

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Examples				Comparative Example	
8	9	10	11	2	3
Immediately after the operation				80.7	65.1
Particulate removing ratio (%)					
76.4	82.7	71.0	73.6		
Pressure loss (mmH ₂ O)				240	590
40	50	55	50		
One hour after the operation					
Particulate removing ratio (%)				52.4	31.6
76.4	82.8	71.3	73.6		
Pressure loss (mmH ₂ O)					
40	50	55	50	3280	6130
24 hours after the operation					
Particulate removing ratio (%)					
77.0	82.5	71.0	73.8	--*	--*
Pressure loss (mmH ₂ O)				--*	--*
40	50	55	50		

*: The combustion of fuel was stopped due to increase in pressure loss.

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As apparent from Table 2, in Examples 8 to 11, unreacted particulates did not remain and were not accumulated and, therefore, the particulate removing ratios and the pressure loss data did not undergo a substantial change. In Comparative Examples 2, 3, unreacted particulates remained and were accumulated one hour after start of the operation of the engine. Accordingly, the pressure loss was considerably increased, causing the fuel combustion in the engine to be stopped.

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Claims

1. A particulate removing catalyst filter in which an oxidizing catalyst is carried on the surface portion of an air-permeable substrate through which exhaust gas is adapted to be passed.
2. A particulate removing catalyst filter according to Claim 1, wherein the air-permeable substrate is

made in the form of a honeycomb structure of which openings at the exhaust gas inlet side and openings at the exhaust gas outlet side are alternately closed such that the exhaust gas is passed through partition walls of said honeycomb structure, and the oxidizing catalyst is carried on the surface portions of said partition walls.

5 3. A particulate removing catalyst filter according to Claim 2, wherein the oxidizing catalyst is carried on one or both sides of the partition walls.

4. A particulate removing method comprising the step of forcibly passing the exhaust gas from a diesel engine through at least one catalyst filter as set forth in Claim 1, said filter being disposed in a casing.

10 5. A particulate removing catalyst filter in which an oxidizing catalyst is carried on a thin wire net or metallic plate having a plurality of fine through-holes.

6. A particulate removing catalyst filter according to Claim 5, wherein the diameter of each of the fine holes is 30 μm or more.

7. A particulate removing catalyst filter according to Claim 6, wherein the diameter of each of the fine holes is in a range from 100 to 500 μm .

15 8. A particulate removing catalyst filter according to Claim 5, wherein the thickness of the wire net or metallic plate is in a range from 0.05 to 0.5 mm.

9. A particulate removing catalyst filter according to Claim 5, wherein the oxidizing catalyst is a layer of metal oxide formed by oxidizing the surface of the wire net or metallic plate.

20 10. A particulate removing catalyst filter according to Claim 5, wherein the wire net or metallic plate is made of stainless steel.

11. A particulate removing method comprising the step of forcibly passing the exhaust gas from a diesel engine through at least one catalyst filter as set forth in Claim 5, said filter being disposed in a casing.

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Fig.1

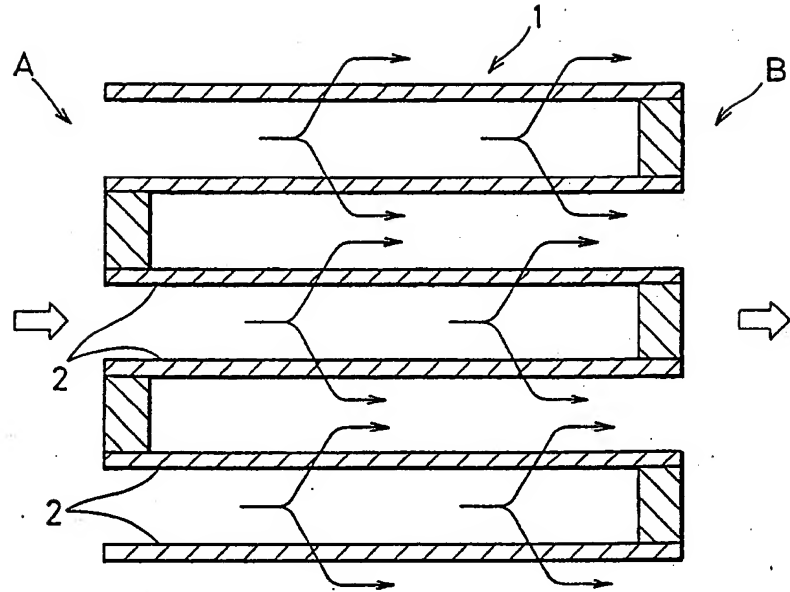


Fig.2

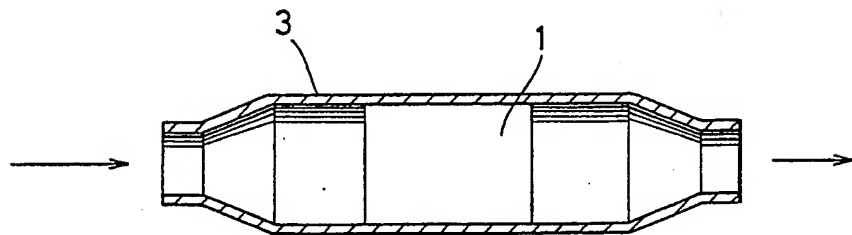


Fig.3

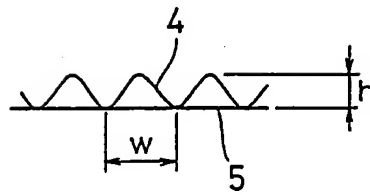


Fig.4

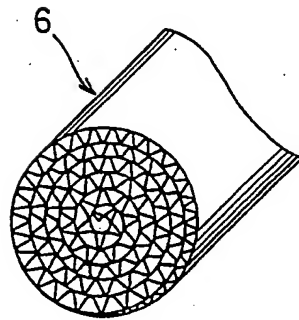


Fig.5

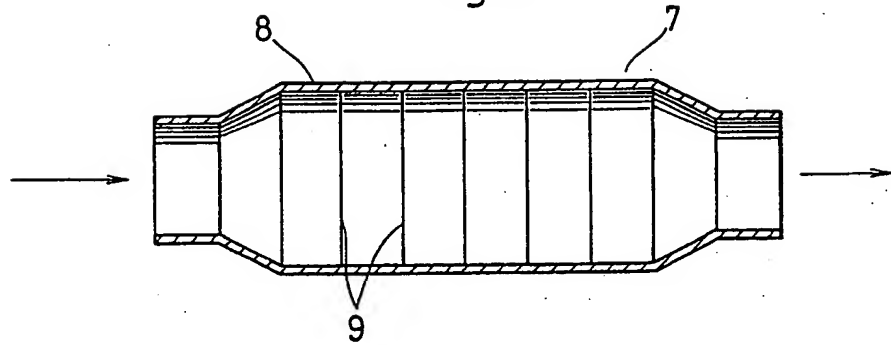


Fig.6

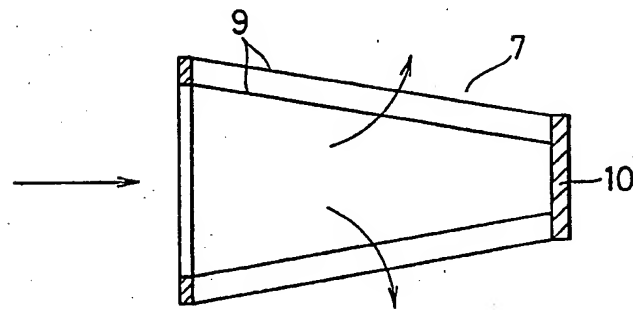


Fig.7

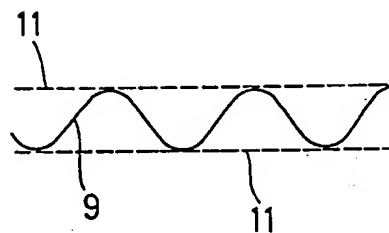


Fig.8

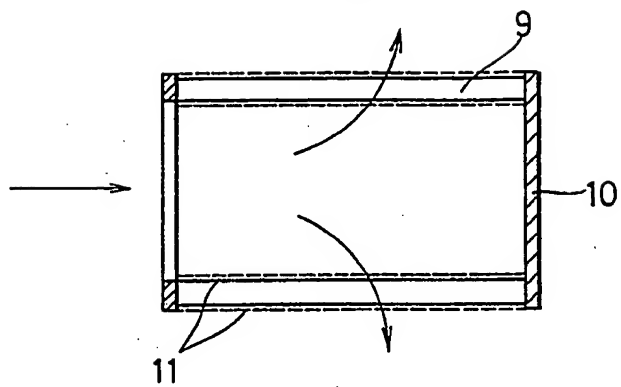


Fig.9

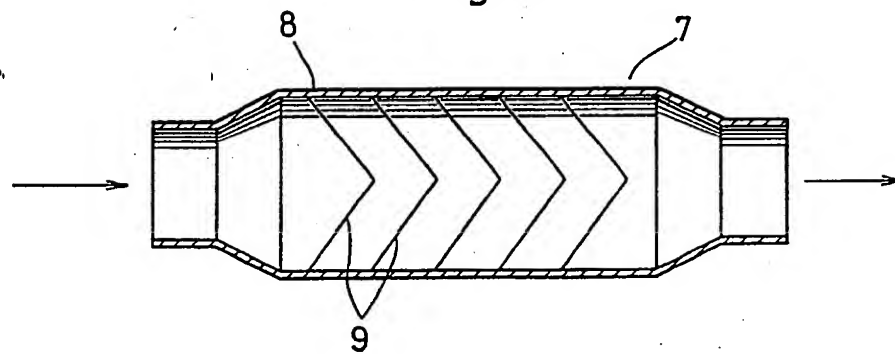
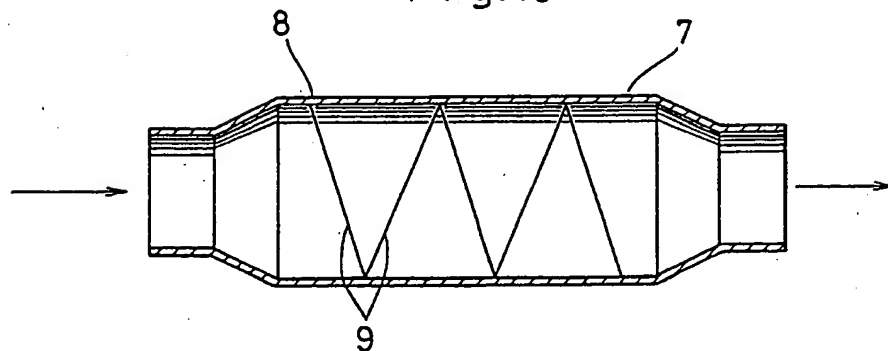


Fig.10





European Patent
Office

EUROPEAN SEARCH REPORT

Application Number

EP 89 11 8876

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
X	PATENT ABSTRACTS OF JAPAN, vol. 9, no. 80 (M-370)[1803], 10th April 1985; & JP-A-59 211 708 (NISSAN JIDOSHA K.K.) 30-11-1984 * Abstract * ---	1-3	F 01 N 3/28 B 01 D 53/36 F 01 N 3/02
X	DE-U-8 323 892 (INTERATOM GmbH) * Abstract * ---	1,2,4,5 10,11	
X	EP-A-0 035 053 (DEGUSSA AG) * Claims 1,5,6,7,12,15; page 5, lines 25-30; page 7, lines 13-36 * ---	1-11	
A	DE-A-3 337 903 (G. REIF) -----		
			TECHNICAL FIELDS SEARCHED (Int. Cl.5)
			B 01 D 53/00 F 01 N 3/00
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 25-01-1990	Examiner POLESAK, H.F.
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

EPO FORM 1503 (03.82) (P0401)

INTERNATIONAL SEARCH REPORT

International application No.

PCT/KR2004/000856

A. CLASSIFICATION OF SUBJECT MATTER

IPC7 B01D 39/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC7 B01D 39/00

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

KIPASS, USP, PAJ "FILTER, DIESEL, MATAL, CATALYST, OXIDE"

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 369163 A1 (SAKAI CHEMICAL INDUSTRY CO., LTD.) 23 May 1990 See the whole document	1-21
A	US 4,973,435 A (ALCAN INTERNATIONAL LIMITED) 27 November 1990 See the abstract and column2 line11 - column6 line46	1-21
A	JP 9-52908 A (IDEMITSUKOSAN CO., LTD.) 25 February 1997 See the abstract and claims1-6	1-21
A	JP 2002-119809 A (TOYOTA CENTRAL RES & DEV LAB INC.) 23 April 2002 See the whole document	1-21
A	JP 1-199613 A (CATALERKOGYO KK) 11 August 1989 See the whole document	1-21

☐ Further documents are listed in the continuation of Box C.☒ See patent family annex.

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"&" document member of the same patent family

Date of the actual completion of the international search

30 JULY 2004 (30.07.2004)

Date of mailing of the international search report

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Name and mailing address of the ISA/KR

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INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No.

PCT/KR2004/000856

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JP 1-199613 A	11.08.1989	NONE	